MINISTRY OF PUBLIC HEALTH OF UKRAINE
Department of human resources policy, education and science
Testing Board

Test items for licensing examination

Krok 1
PHARMACY
General Instruction

Every one of these numbered questions or unfinished statements in this chapter corresponds to answers or statements endings. Choose the answer (finished statements) that fits best and fill in the circle with the corresponding Latin letter on the answer sheet.


The book includes test items for licensing integrated examination “Krok 1. Pharmacy” and further use in teaching.

The book has been developed for students of pharmaceutical faculties and academic staff of higher medical educational establishments.


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1. Calculation of thermal effects of chemical reactions at pharmaceutical production is based on the Hess law. This law states that thermal effect of a reaction is determined by:

A. Initial and final states of system  
B. Mechanism of chemical change  
C. Route of chemical change  
D. Number of intermediate stages  
E. Process duration

2. A chemical laboratory received a drug that is a mixture of glucose and mannose. To identify these substances in the mixture the following method can be applied:

A. Thin-layer sorbent chromatography  
B. Polarimetry  
C. Spectrophotometry  
D. Polarography  
E. Amperometric titration

3. Microscopy of leaf epidermis of Lamiaceae (Labiatae) family plants revealed that both accessory cells are perpendicular to a stoma. Such stomata are called:

A. Diacytic  
B. Paracytic  
C. Anisocytic  
D. Anomocytic  
E. Tetracytic

4. Stem thickening occurs due to functioning of the following structures:

A. Lateral meristem  
B. Apical meristem  
C. Wound meristem  
D. Intercalary meristem  
E. Endoderm

5. A patient has developed anuria. Blood pressure is 50/20 mm Hg. What process of uropoiesis caused acute decrease of urination?

A. Glomerular filtration  
B. Obligate reabsorption  
C. Facultative reabsorption  
D. Tubular secretion  
E. -

6. In snake venom there is a substance that causes erythrocyte hemolysis, when it is introduced into a human organism. Blood test revealed a large amount of lysolecithin (lysophosphatidylcholine). What enzyme leads to accumulation of lysolecithin in blood?

A. Phospholipase A2  
B. Phospholipase A1  
C. Phospholipase C  
D. Phospholipase D  
E. Neuraminidase

7. Sulfanilamide drugs contain primary aromatic amides in their structure. Specify the method of quantitative determination of these compounds:

A. Nitritometry  
B. Iodometry  
C. Dichromatometry  
D. Permanganatometry  
E. Cerimetry

8. Microbe survival within environment is facilitated by spore formation. What microorganisms of those listed below are spore formers:

A. Clostridia  
B. Bacteroides  
C. Staphylococci  
D. Peptococci  
E. Peptostreptococci

9. Dissociation degree in 0,01 M water solution is the same for all the strong electrolytes listed below. Name the substance with the highest boiling temperature:

A. Al₂(SO₄)₃  
B. KCl  
C. Na₃PO₄  
D. Cu(NO₃)₂  
E. K₃PO₄

10. Proteins are of great importance
for vital functions. What value of pH results in zero electrophoretic mobility of gelatin (gelatin isoelectric point equals 4.7)?

A. 4.7
B. 7.0
C. 14.0
D. 5.5
E. 9.4

11. Enzymes are widely used as drugs in pharmacy. What is the main difference that separates enzymes from non-biological catalysts?

A. High specificity and selectivity
B. High universality
C. Low universality
D. High dispersion
E. High homogeneity

12. A patient with chronic constipation had been prescribed bisacodyl. After 3 weeks of treatment the patient noticed a reduction of laxative effect. This is caused by the development of the following side-effect:

A. Tolerance
B. Dependence
C. Sensibilization
D. Cumulation
E. Disbacteriosis

13. Solution under analysis received chloroform and, drop by drop, chlorine water. Chloroform layer colored orange, which indicates the presence of:

A. Bromide ions
B. Iodide ions
C. Sulfite ions
D. Sulfate ions
E. Nitrate ions

14. A species of Ericaceae family is characterized by the following type of leaves: alternate leaf arrangement, short footstalk, leathery, elliptic or obovate with retuse tip, downturned edges; upper surface is dark-green, lower surface is light-green with punctate glandules. Name this species:

A. Vaccinium vitis-idaea
B. Arctostaphilos uva-ursi
C. Vaccinium oxyccoccus
D. Vaccinium myrtillus
E. Ledum palustre

15. In an emergency situation a scuba diver has quickly risen from the depth to the surface in violation of safety regulations. He is unconscious, presents with respiratory failure and cardiac activity disorder as the result of decompression sickness. What complication can develop in the scuba diver?

A. Gas embolism
B. Fat embolism
C. Air embolism
D. Cellular embolism
E. Thromboembolism

16. Parents of a 10-year-old child have made an appointment with endocrinologist due to complaints of the child’s low height. The child’s appearance is corresponding with that of a 5-year-old. What hormone causes such changes in physical development, if its secretion is disrupted?

A. Somatotropic hormone
B. Adrenocorticotropic hormone
C. Thyroxin
D. Testosterone
E. Insulin

17. A patient complains of tachycardia, insomnia, weight loss, irritability, sweating. Objectively: the patient has goiter and slight exophthalmos. What gland is affected, and what functional disorder is it?

A. Hyperthyroidism
B. Hypothyroidism
C. Hyperparathyroidism
D. Hypoparathyroidism
E. Adrenomedullary hyperfunction

18. In titration analysis aimed at determining the substances by means of mercurimetry the following substance can be used as the indi-
A. Diphenylcarbazide
B. Potassium chromate
C. Eriochrome black T
D. Starch
E. Tropeolin OO

19. A patient has been hospitalised with pneumonia. What kind of respiratory failure does the patient have?
A. Restrictive
B. Obstructive
C. Central
D. Peripheral
E. Thoracic diaphragm

20. In the state of fright the following signs can be observed: acute pallor of face, tremor of extremities. What kind of ischemia can be observed in such a condition?
A. Angiospastic
B. Compression
C. Obstructive (thrombus)
D. Metabolic
E. Obstructive (vascular wall thickening)

21. What are the indications for the use of naloxone?
A. Acute intoxication with narcotic analgesics
B. Heavy metals intoxication
C. Intoxication with cardiac glycosides
D. Intoxication with ergot alkaloids
E. Atropine sulfate intoxication

22. Thermodynamic calculations allow us to determine the possibility and direction of spontaneous processes. In an isolated system the change of the following thermodynamic function is used for this purpose:
A. Entropy
B. Gibbs energy
C. Helmholtz energy
D. Internal energy
E. Enthalpy

23. What drug should be administered for individual prevention of malaria?
A. Chingamin
B. Rifampicin
C. Ampicillin
D. Gentamicin
E. Biseptol (Co-Trimoxazole)

24. A sample section of an axial body shows a complex consisting of phellogen and its derivatives - cork and phelloderm. This tissue is called:
A. Periderm
B. Colenchyma
C. Sclerenchyma
D. Epiblema
E. Epidermis

25. The Mohr method is used to determine mass concentration of sodium chloride in an isotonic solution. Titration is carried out with the following indicator:
A. Potassium chromate
B. Fluorescein
C. Ammonium iron (III) sulfate
D. Diphenylcarbazone
E. Ferroin

26. An injured person exhibits the following signs at the site of trauma: skin redness, throbbing small arteries, elevated local temperature, increased tissue turgor. What local blood circulation disorder are these presentations typical of?
A. Arterial hyperemia
B. Venous hyperemia
C. Thrombosis
D. Embolism
E. Ischemia

27. Racemose clusters of calcium carbonate crystals are detected among the waste products of a protoplast. These crystals are:
A. Cystoliths
B. Isolated crystals
C. Raphides
D. Styloids
E. Druses

28. Purine ring biosynthesis occurs in ribose-5-phosphate by gradual
accumulation of nitrogen and carbon atoms and closing the rings. The source of ribose phosphate is the process of:

A. Pentose phosphate cycle  
B. Glycolysis  
C. Glyconeogenesis  
D. Gluconeogenesis  
E. Glycogenolysis

29. Number of freedom degrees at the point of intersection of liquidus with Y-axis on the fusibility chart of a two-component system would equal:

A. 0  
B. 2  
C. 3  
D. 4  
E. 1

30. The fourth group of cations includes the following cations: $Al^{3+}$, $Sn^{2+}$, $Sn(IV)$, $As(V)$, $As(III)$, $Zn^{2+}, C^{3+}$. The group reagent for the fourth group of cations is the solution of:

A. $NaOH, H_2O_2$  
B. $HCl$  
C. $NH_3, H_2O_2$  
D. $H_2C_2O_4$  
E. $H_2SO_4, H_2O_2$

31. In the qualitative analysis that involves precipitation of sulfates of the third analytical group cations ($Ca^{2+}, Sr^{2+}, Ba^{2+}$) the solubility of sulfates can be reduced by adding:

A. Ethyl alcohol  
B. Distilled water  
C. Benzene  
D. Chloroform  
E. Amyl alcohol

32. Reaction rate constant values allow to draw conclusions regarding processes of synthesis of various drugs. What factor affects reaction rate constant?

A. Temperature  
B. Pressure  
C. Volume  
D. Concentration  
E. Reaction time

33. Uric acid is a derivative of:

![Uric acid structure](image)

A. Purine  
B. Indole  
C. Pyrazine  
D. Pyrazole  
E. Pyridine

34. In terms of water-air interface the following substance is a surfactant:

A. Valeric acid  
B. $HCl$  
C. $NaOH$  
D. Urea  
E. -

35. At the sixth month of pregnancy a woman has been diagnosed with severe iron-deficiency anemia. Appearance of the following elements in her blood became the diagnostic character:

A. Hypochromic erythrocytes  
B. Macrocytes  
C. Megalocytes  
D. Reticulocytes  
E. Erythroblasts

36. What compound produces phthalic acid during oxidation?

![Phthalic acid structure](image)
37. To distinguish between phenol and salicylic acid the following reagent is used:

A. Sodium bicarbonate solution
B. Iron (III) chloride solution
C. Sodium hydroxide solution
D. Sodium chloride solution
E. Bromine solution

38. Nitrite ions in presence of nitrate ions can be detected with:

A. Crystalline antipyrine in presence of diluted HCl
B. Crystalline sodium thiosulfate
C. Dimethylglyoxime
D. Crystalline iron (III) sulfate
E. Diphenylcarbazone

39. Analysis of the cerebrospinal fluid of a child with signs of purulent lesion of brain tunics revealed gram-negative bean-shaped diplococci. What provisional diagnosis can be made based on the analysis results?

A. Meningitis
B. Gonorrhea
C. Cholera
D. Plague
E. Anthrax

40. Aqueous solution of CaCl₂ with 10% mass concentration is used for intravenous injections. What is the maximum value of isotonic coefficient of CaCl₂ in an aqueous solution?

A. 3
B. 4
C. 2
D. 5
E. 1

41. Which of these reactions can be used to identify the primary amino group?
A. 

\[ \text{H}_3\text{C} - \text{CH}_2 - \text{NH}_2 + \text{CHCl}_3, \text{KOH} \]

\[ \text{H}_3\text{C} - \text{CH}_2 - \text{NH}^- + \text{KCl} + \text{H}_2\text{O} \]

B. 

\[ \text{H}_3\text{C} - \text{CH}_2 - \text{NH}_2 + \text{H}_3\text{C} - \text{I} \]

\[ \text{H}_3\text{C} - \text{CH}_2 - \text{NH}^+ - \text{CH}_3 + \text{HI} \]

C. 

\[ \text{H}_3\text{C} - \text{CH}_2 - \text{NH}_2 + (\text{CH}_3\text{CO})_2\text{O} \]

\[ \text{H}_3\text{C} - \text{CH}_2 - \text{NH}^- - \text{CH}_3 + \text{CH}_3\text{COOH} \]

D. 

\[ \text{H}_3\text{C} - \text{CH}_2 - \text{NH}_2 + \text{HCl} \]

\[ \text{H}_3\text{C} - \text{CH}_2 - \text{NH}_3^+ - \text{Cl}^- \]

E. 

\[ \text{H}_3\text{C} - \text{CH}_2 - \text{NH}_2 \]

\[ \text{H}_3\text{C} - \text{CH}_2 - \text{NO}_2 \]

42. Bacterial culture obtained from patient \textbf{DOES NOT} grow when exposed to oxygen. Conditions suitable for bacterial culture growth can be created in:

A. Anaerobic culture jar  
B. Serum-supplemented medium  
C. Pasteur oven  
D. Krotov apparatus  
E. Oxidative medium

43. In the course of long-term treatment of an infectious patient with penicillin, the pathogen transformed into the L-form. What changes occur in the pathogen cell in case of L-transformation?

A. Absence of a cell wall  
B. Absence of flagella  
C. Absence of a capsule  
D. Absence of a spore  
E. Absence of inclusions

44. Sedimentation analysis has been applied for assessment of air purity in an aseptic unit of a pharmacy. The test resulted in growth of the small colonies with areas of hemolysis. What medium was used for inoculation?

A. Blood agar  
B. Levine’s agar (Eosin Methylene Blue agar)  
C. Endo agar  
D. Ploskirev’s agar  
E. Egg-yolk salt agar

45. What enzyme allows for synthesis of various genes from template-RNA to DNA in genetic engineering (this enzyme catalyzes the process discovered in RNA-viruses)?

A. Reverse transcriptase  
B. Exonuclease  
C. DNA-ligase  
D. Helicase  
E. Endonuclease

46. Smears from tonsillar coating of a patient were stained by Neisser’s method. Microscopy revealed thin yellow V-shaped bacilli with dark-blue grains at their ends. Make the preliminary diagnosis:

A. Diphtheria  
B. Measles  
C. Tuberculosis  
D. Whooping cough  
E. Influenza

47. According to van’t Hoff rule, when the temperature is raised by 10 degrees, the reaction rate increases by:

A. 2-4 times  
B. 1.5 times  
C. 5 times  
D. 10 times  
E. Temperature does not affect reaction rate

48. Potassium dichromate solution was added into a solution obtained after the precipitate consisting of group II chloride cations was processed with hot water. Yellow
precipitate was produced; the precipitate is insoluble in acetic acid, but soluble in alkali. What cations were present in the solution under investigation?

A. Lead (II)  
B. Mercury (II)  
C. Barium  
D. Silver (I)  
E. Calcium

49. Diet of an individual must contain vitamins. What vitamin is usually prescribed for treatment and prevention of pellagra?

A. Vitamin \( \text{PP} \)  
B. Vitamin \( C \)  
C. Vitamin \( A \)  
D. Vitamin \( B_1 \)  
E. Vitamin \( D \)

50. Thiocyanatometric titration method requires secondary standard solution of potassium thiocyanate. This solution is standardized with standard solution of:

A. Silver nitrate  
B. Hydrochloric acid  
C. Sulfuric acid  
D. Iron (II) sulfate  
E. Copper (II) nitrate

51. To identify a drug by means of thin-layer chromatography the following parameter is used:

A. \( R_f \)  
B. \( n \)  
C. \( E, mV \)  
D. \( I, A \)  
E. \( K_p \)

52. Intracellular metabolism of glycerol starts with its activation. What compound is formed in the first reaction of its conversion?

A. \( \alpha \)-glycerolophosphate  
B. Pyruvate  
C. Lactate  
D. Choline  
E. Acetyl coenzyme A

53. According to the Pharmacopoeia regulations non-sterile drugs can contain certain microorganisms. Name the microorganisms that \textbf{CANNOT} be present:

A. Enterobacteriaceae  
B. Yeast fungi  
C. Micrococci  
D. Mold fungi  
E. Sarcinae

54. According to IUPAC nomenclature the following is the name of nicotinic acid:

![Nicotinic Acid](image)

A. Pyridine-3-carboxylic acid  
B. Pyridine-2-carboxylic acid  
C. Pyridine-4-carboxylic acid  
D. 4-carboxypyridine  
E. 2-carboxypyridine

55. Select a name that corresponds with the formula: \( CH_3 - C \equiv N \):

A. Acetic acid nitrile  
B. Acetamide  
C. Acetic anhydride  
D. Acetoxime  
E. Ethyl isocyanide

56. Tryptophan amino acid is a derivative of:

![Tryptophan](image)

A. Indole  
B. Coumarin  
C. Pyridine  
D. Imidazole  
E. Purine

57. Flax seeds are used in medicine as coating agents, due to the following ability of their secondary membranes:
A. Sliming
B. Suberization
C. Gummosis
D. Lignification
E. Mineralization

58. Which of the following reactions is required to obtain an azo dye out of an aromatic amine?

A. Diazotization and azo compound
B. Reduction and diazotization
C. Diazotization and interaction with potassium cyanide
D. Salt formation and nitration
E. Alkylation and nitrosation

59. How many asymmetric carbon atoms and stereoisomers are there in tartaric acid?

A. Two asymmetric atoms and three stereoisomers
B. One asymmetric atom and two stereoisomers
C. Two asymmetric atoms and four stereoisomers
D. No asymmetric atoms and no stereoisomers
E. Two asymmetric atoms and two stereoisomers

60. Chemically, ethers are quite inert compounds. Éthers decompose even at a room temperature under the effect of the following haloid acid:

A. HI
B. HBr
C. HCl
D. HF
E. HClO

61. Introduction of an electron acceptor substitute into molecule is known to increase acid strength. What substance demonstrates the most explicit acidic properties?

A. Trichloracetic acid
B. Dichloroacetic acid
C. Chloroacetic acid
D. Acetic acid
E. Propionic acid

62. Reaction of benzene sulfonation produces:

A. 

B. 

C. 

D. 

E. 

63. Select the formula of diazonium salt:
A. \[
\begin{array}{c}
\text{CH}_2 - N = N - \text{Br} \quad \text{N}
\end{array}
\]
B. \[
\begin{array}{c}
\text{N} - \text{N} - \text{OH}
\end{array}
\]
C. \[C_6H_5 - N = O\]
D. \[C_6H_5 - NH - C(O) - CH_3\]
E. \[(CH_3)_2N - N = O\]

64. A pharmaceutical enterprise produces a tetanus-specific preventive drug. Which drug of those listed below is it?
A. Anatoxin
B. Dead vaccine
C. Live vaccine
D. Immunoglobulin
E. Recombinant vaccine

65. Inoculation in a nutrient medium was performed to determine probable contamination of a drug with fungi. The colonies are large, resembling sour cream. What nutrient medium had been used in this case?
A. Sabouraud
B. Loewenstein-Jensen medium
C. Roux
D. Loeffler
E. FINN-II

66. Aniline can be converted into the water-soluble salt, if processed with the solution of:
A. Hydrochloric acid
B. Sodium hydroxide
C. Sodium sulfate
D. Ethanol
E. Dimethylamine

67. Specify the reagent necessary for the following transformation:
A. \[NH_2NH_2\]
B. \[NH_2OH\]
C. \[CH_3NH_2\]
D. \[C_6H_5NHNH_2\]
E. \[NH_3\]

68. Select the compound with amphoteric properties (which reacts both with acids and bases and produces salts):
A. \[
\begin{array}{c}
\text{N}
\end{array}
\]
B. \[
\begin{array}{c}
\text{N}
\end{array}
\]
C. \[
\begin{array}{c}
\text{N}
\end{array}
\]
D. \[
\begin{array}{c}
\text{N}
\end{array}
\]
E. \[
\begin{array}{c}
\text{N}
\end{array}
\]

69. Select the formula of pentene-2 from the list:
A. \[CH_3 - CH_2 - CH = CH - CH_3\]
B. \[CH_3 - CH_2 - CH_2 - CH_2 - CH_3\]
C. \[CH_3 - CH_2 - CH_2 - CH = CH_2\]
D. \[CH_3 - CH_2 - CH_2 - CH_3\]
E. \[CH_3 - CH = CH - CH_3\]

70. The end product of starch hydrolysis is:
A. D-glucose
B. D-fructose
C. Saccharose
D. Maltose
E. D-galactose
71. Determine the compounds of $X$ and $Y$ in the reaction:

$$\text{CH}_2=\text{CH}_2 + \text{Br}_2 \xrightarrow{\text{Br}_2} X \xrightarrow{\text{NaOH, alcoholic solution}} Y$$

A. $\text{CH}_2=\text{CH}_2$ and $\text{CH}≡\text{CH}$
B. $\text{CH}_3=\text{CH}_2$ and $\text{CH}_2=\text{CH}_2$
C. $\text{CH}_3=\text{CHBr}_2$ and $\text{HC}≡\text{CH}$
D. $\text{CH}_3=\text{CHBr}$ and $\text{CH}_2=\text{CH}_2$
E. $\text{CH}_2=\text{CHBr}$ and $\text{CH}_3=\text{C}=\text{O}$

72. What reagent can help distinguish between starch and glucose?

A. $I_2$
B. $Br_2$
C. $KMnO_4$
D. $K_2Cr_2O_7$
E. $FeCl_3$

73. Select the formula, where carbon atoms numbering complies with IUPAC replacement nomenclature:

74. What reagent allows to simultaneously detect the presence of both aldehyde group and glycol fragment in glucose molecule?

A. $Cu(OH)_2$
B. $Br_2$
C. $AlCl_3$
D. $FeCl_3$
E. $KMnO_4$

75. What class of organic compounds is characterized by the presence of $C ≡ N$ group?

A. Nitriles
B. Amines
C. Nitro compounds
D. Alcohols
E. Aldehydes
76. Accidental ingestion of death cap mushrooms containing α-amanitin causes intoxication. What enzyme is inhibited with this toxine?

A. RNA polymerase II  
B. DNA polymerase  
C. DNA synthetase  
D. Peptidyl transferase  
E. Translocase

77. Specify the substance that results from the following reaction:

$$CH \equiv CH \xrightarrow{HOH, Hg^{2+}} ?$$

A. Ethanal  
B. Ethanol  
C. Propanal  
D. Propanone  
E. Acetic acid

78. Choose the indicator and titration method to determine hydrogen carbonate ions in a drug:

A. Methyl-orange, acidimetry  
B. Phenolphthalein, acidimetry  
C. Methyl-orange, alkalimetry  
D. Phenolphthalein, alkalimetry  
E. Murexide, acidimetry

79. What titrant is used in bromatometry?

A. $K BrO_3$  
B. $K Br$  
C. $Br_2$  
D. $K BrO_4 + KCl$  
E. $K BrO_4$

80. If the amount of a high-molecular substance added to the sol is very small, it can not increase but decrease its stability. This phenomenon is called:

A. Sensibilization  
B. Solubilization  
C. Mutual coagulation  
D. Colloidal protection  
E. Sol adaptation

81. An ophthalmologist has detected increased time of dark adaptation in a patient. What vitamin deficiency can result in such symptom?

A. A  
B. C  
C. K  
D. $B_1$  
E. $B_6$

82. A 70-year-old patient presents with cardiac and cerebral atherosclerosis. Examination revealed changes of blood lipid spectre. Increase of the following lipoproteins plays a significant role in atherosclerosis pathogenesis:

A. Low-density lipoproteins  
B. Very low-density lipoproteins  
C. Intermediate density lipoproteins  
D. High-density lipoproteins  
E. Chylomicrons

83. During containment measures following Chornobyl Nuclear Power Plant disaster a worker has been exposed to a dose of ionizing emission of 6 Gy (600 R). The worker complains of general fatigue, nausea, dizziness, labile blood pressure and heart rate, short-term leukocytosis with lymphopenia. What stage of acute radiation sickness can be characterized by such presentations?

A. Prodromal  
B. Manifest  
C. Latent  
D. Recovery  
E. Long-term consequences

84. During influenza epidemic a patient with severe case of disease developed hacking cough and chest pain; signs of focal pneumonia were visible on X-ray. Microscopy of sputum detected large number of pneumococci. What type of infection is it?

A. Secondary  
B. Superinfection  
C. Abortive  
D. Relapse  
E. Reinfection

85. A factory producing typhoid fever vaccine cultivates bacteria of virulent strain in optimal nutrient medium.
Then the cells are separated from culture fluid by means of centrifugation and processed with formalin. What type of vaccine is it?

A. Inactivated
B. Attenuated
C. Chemical
D. Anatoxin
E. Autovaccine

86. Spore and pollen analysis revealed tetrahedral spores with a semicircular base and reticular surface in the pollen. It is the pollen of:

A. Lycopodiophyta
B. Equisetiphyta
C. Bryophyta
D. Polypodiophyta
E. Pinophyta

87. A patient demonstrates milky-white color of blood plasma due to high content of chylomicrons. Disintegration of triacylglycerol is disrupted. Deficiency of the following enzyme activity is observed:

A. Lipoprotein lipase
B. Amylase
C. Tripsin
D. Cholesterol esterase
E. Lactase

88. A woman noticed that a cut on her skin was still bleeding even after 20 minutes had passed. What vitamin deficiency causes such condition?

A. Vitamin K
B. Vitamin A
C. Vitamin D
D. Vitamin E
E. Vitamin B₁₂

89. Primary structure of nucleic acids is a polynucleotide chain that has a certain composition and order of the nucleotides. What bonds stabilize this structure?

A. 3′, 5′-phosphodiester
B. Peptide
C. Glycosidic
D. Disulfide
E. Amide

90. *Quercus robur* leaves have the following type of lamina shape and division:

A. Pinnatilobate
B. Trilobate
C. Pinnatifipartite
D. Palmatilobate
E. Palmatipartite

91. Natural peptides can perform various functions. What bioactive peptide is a major antioxidant and performs coenzyme functions?

A. Glutathione
B. Bradykinin
C. Oxytocin
D. Liberin
E. Anserine

92. Microbiological purity of tableted drugs has been tested at a factory. Samples cultivation in mannitol salt agar resulted in growth of golden-yellow colonies, microscopic examination of colonies detected gram-positive globular bacteria positioned in clusters; microorganisms had plasma coagulation properties. What pure bacterial culture was obtained?

A. Staphylococcus aureus
B. Enterobacteriaceae
C. Staphylococcus epidermidis
D. Staphylococcus saprophyticus
E. Pseudomonas aeruginosa

93. Mosaic discoloration of leaves has been detected at medicinal plantations. What microorganisms are the cause of such damage?

A. Plant-pathogenic viruses
B. Plant-pathogenic bacteria
C. Plant-pathogenic fungi
D. Protozoa
E. Rickettsia

94. Contrykal is used to prevent pancreatic autolysis. This drug is the
inhibitor of the following enzymes:

A. Proteases
B. Lipases
C. Glycosidases
D. Nucleases
E. Synthetases

95. A dry-heat box is used for sterilization of various materials and instruments in a bacteriological laboratory. This sterilization method can be applied to the following objects:

A. Glass test tubes
B. Rubber gloves
C. Simple nutrient medium
D. Wire inoculating loops
E. Physiological solution

96. A sample of water used in drug production has been sent to a laboratory for sanitary and virological analysis. Presence of what virus group will be indicative of faecal contamination of water and, thus, the need for its additional purification?

A. Picornaviridae
B. Herpesviridae
C. Orthomyxoviridae
D. Retroviridae
E. Flaviviridae

97. In potentiometric titration the following indicator electrode is used for quantitative determination of chloride and borate acids in their mixture:

A. Glass
B. Silver-chlorine
C. Silver
D. Platinum
E. Calomel

98. An elderly man exhibits low levels of red blood cells and hemoglobin in blood; however, his color index is 1.3. Blood smear analysis revealed megaloblasts. What type of anemia is observed in this case?

A. $B_{12}$-folic acid deficiency
B. Iron-deficiency
C. Acquired hemolytic
D. Hereditary hemolytic
E. Chronic posthemorrhagic

99. Morphological analysis of poplar inflorescence shows that it is a simple monopodial inflorescence: main axis is drooping, the flowers are sessile, unisexual. Specify the type of inflorescence:

A. Catkin
B. Head
C. Capitulum
D. Cyme
E. Panicle

100. After drinking milk a 1-year-old child developed diarrhea, flatulence. The baby is likely to have deficiency of the following enzyme:

A. Lactase
B. Maltase
C. Aldolase
D. Hexokinase
E. Glycosidase

101. Patients with severe depression demonstrate decreased serotonin levels in brain and cerebrospinal fluid. What aminoacid is a serotonin precursor?

A. Tryptophan
B. Threonine
C. Tyrosine
D. Glutamic acid
E. Aspartic acid

102. During reaction of silver cations identification first $HCl$ and then ammonia solution have been added to the solution. What compound was produced as the result?

A. $[Ag(NH_3)_2]Cl$
B. $[Ag_2(NH_3)_3]Cl$
C. $AgOH$
D. $AgCl$
E. $[Ag(NH_3)_3]Cl$

103. Microscopic study of soy bean seeds stained with Sudan III revealed
droplets of various sizes. These droplets are:

A. Lipids  
B. Proteins  
C. Starch  
D. Inulin  
E. Glycogen

104. In a chemical analytical laboratory a chemist investigates a solution of anion mixture. When antipyrin solution is added it colors emerald-green. This analytical effect signifies presence of the following anions:

A. Nitrite  
B. Nitrate  
C. Acetate  
D. Tartrate  
E. Citrate

105. Selective solvents are used in laboratories and factories to isolate and refine essential oils, alkaloids, antibiotics and other pharmaceutical substances. This process is called:

A. Extraction  
B. Sedimentation  
C. Coagulation  
D. Flocculation  
E. Flotation

106. When preparing a solution, an analytical pharmacist converted a freshly formed precipitate into a sol by treating it with an electrolyte solution. What method of obtaining disperse systems was used by the pharmacist?

A. Peptization  
B. Physical condensation  
C. Chemical condensation  
D. Solvent exchange  
E. Condensation from steam

107. Sol of iron (III) hydroxide is positively charged. Specify the ion that has the lowest coagulation threshold:

A. $SO_4^{2-}$  
B. $Cl^-$  
C. $Cu^{2+}$  
D. $Na^+$  
E. $J^-$

108. Fatty acids synthesis occurs in human body. What compound is initial in this process?

A. Acetyl coenzyme A  
B. Vitamin C  
C. Glycine  
D. Succinate  
E. Cholesterol

109. Specify the standard solutions that are used in permangananometry to quantify the oxidants by means of back titration:

A. Potassium permanganate, iron (II) sulfate  
B. Potassium dichromate, sodium thiosulfate  
C. Potassium bromate, sodium thiosulfate  
D. Potassium iodate, sodium thiosulfate  
E. Cerium (IV) sulfate, iron (II) sulfate

110. What indicator is used to fix the endpoint of mercurimetric titration?

A. Thiocyanate complexes of iron (III)  
B. Fluorescein  
C. Eosin  
D. Murexide  
E. Potassium chromate

111. A patient with alcoholic cirrhosis complains of general weakness and dyspnea. The following is revealed: decrease of blood pressure, ascites, dilation of superficial veins of the stomach anterior wall, esophageal varicose veins dilatation, splenomegaly. What hemodynamics disorder does the patient suffer from?
A. Portal hypertension
B. Left ventricular failure
C. Right ventricular failure
D. Cardiac insufficiency
E. Collapse

112. Pharmacy has received viricides. Choose the viricide for influenza treatment from the list given below:
A. Rimantadine
B. Metisazone
C. Levamisole
D. Azidothimidine
E. Acyclovir

113. Colloid silver preparations Protargolum and Collargolum are widely used in medical practice as bactericidal drugs. In addition to the active ingredients, these drugs contain protein compounds. What is the function of proteins in these preparations?
A. Prevention of colloidal solution coagulation
B. Prolongation of shelf-life
C. Reduction of side effects
D. Improvement of drug technology
E. Potentiation of bactericidal action of silver

114. Pharmacies receive large amounts of sterile medical products (dressing, rubber gloves, catheters, etc.). What ensures their sterility during production?
A. Alpha irradiation
B. Beta irradiation
C. Gamma irradiation
D. Infrared irradiation
E. Ultraviolet irradiation

115. What type of conducting bundles is characteristic of all root zones of one-seeded plants?
A. Radical
B. Central phloem (Amphivasal)
C. Central xylem (Amphicribal)
D. Bilateral
E. Collateral

116. The Volhard method is used to define mass concentration of sodium chloride. Name the titrant of this method:
A. Ammonium thiocyanate
B. Mercury (I) nitrate
C. Sodium tetraborate
D. Mercury (II) nitrate
E. Sodium hydroxide

117. What cation of the 4th analytical group is present in a solution, if its reaction with the group reagent results in formation of yellow precipitate?
A. $Cr^{3+}$
B. $Zn^{2+}$
C. $Sn^{2+}$
D. $Al^{3+}$
E. $Sn(IV)$

118. A patient has icteric skin; unconjugated bilirubin content in blood is high; conjugated bilirubin in urine is not detected. There is significant amount of urobilin in urine and stercobilin in feces. Name the pathology characterized by given symptoms:
A. Hemolytic jaundice
B. Obstructive jaundice
C. Jaundice of the newborn
D. Hepatocellular jaundice
E. Atherosclerosis

119. What anion of the 2nd analytic group produces black precipitate with the group reagent $AgNO_3$?
A. $S^{2-}$
B. $I^-$
C. $Cl^-$
D. $Br^-$
E. $NCS^-$

120. You are studying a silvery downy plant of Asteraceae family, which is rich with essential oils and bitters. Harvested are apical sprouts with panicle of small round flower heads. This plant is:
A. Artemisia absinthium  
B. Arctium lappa  
C. Bidens tripartita  
D. Calendula officinalis  
E. Chamomilla recutita

121. A 46-year-old patient was found to have hyperactivity of creatine kinase in the blood serum. What pathology can be suspected?

A. Myocardial infarction  
B. Acute pancreatitis  
C. Chronic hepatitis  
D. Hemolytic anemia  
E. Renal failure

122. Emulsions containing under 0.1% of dispersed phase (in volume) are classified as:

A. Diluted  
B. Concentrated  
C. High-concentration  
D. Water-in-oil type  
E. Oil-in-water type

123. A patient with pulmonary carcinoma has developed a case of pleurisy. Large amount of hemorrhagic exudate was obtained for analysis. What component is specific for hemorrhagic exudate?

A. Erythrocytes  
B. Leukocytes  
C. Platelets  
D. Fibrin  
E. Pus

124. A pharmaceutical enterprise offers wide range of antimicrobial agents. Select the broad spectrum antimicrobial agent:

A. Tetracycline  
B. Rimantadine  
C. Nystatin  
D. Griseofulvin  
E. Phthalazolum

125. Cataract (lenticular opacity) has developed in a 52-year-old woman with diabetes mellitus. Lenticular opacity was caused by intensification of the following processes:

A. Protein glycosylation  
B. Lipolysis  
C. Ketogenesis  
D. Protein proteolysis  
E. Gluconeogenesis

126. Common nettle, hop, and black elderberry require soils rich in nitrogen compounds. Such plants are called:

A. Nitrophyles  
B. Nitrophobes  
C. Calciphiles  
D. Calciphobes  
E. Halophytes

127. A man received a radiation dose of 30 Gy. He presents with necrotic angina, disorders of the gastrointestinal tract. Blood tests revealed anemia, leukopenia and thrombocytopenia. What stage of acute radiation sickness is observed in the patient?

A. Manifest illness stage  
B. Prodromal stage  
C. Latent stage  
D. Recovery  
E. -

128. A patient with hyperproduction of thyroid hormones has been prescribed Merkazolilum. This drug inhibits the following enzyme participating in iodothyronine synthesis:

A. Iodide peroxidase  
B. Aromatase  
C. Reductase  
D. Decarboxylase  
E. Aminotransferase

129. A patient with croupous pneumonia presents with sharp increase of body temperature up to 39°C, which persisted for 9 days with daily amplitude of 1 degree. What temperature curve could be observed?
130. Employees of an enterprise were vaccinated with “Influvac” for specific prevention of influenza. What type of immunity will develop in those vaccinated?

A. Artificial active
B. Innate congenital
C. Artificial passive
D. Natural active
E. Natural passive

131. Soil microflora often includes representatives of pathogenic microorganisms. Specify the diseases with causative agents that remain viable in the soil for a long time:

A. Tetanus and gas anaerobic infection
B. Tuberculosis and mycobacterioses
C. Colibacillosis and cholera
D. Leptospirosis and plague
E. Typhoid fever and dysentery

132. Removal of low-molecular impurities from colloidal systems and high-molecular compound solutions by means of semipermeable membrane diffusion is called:

A. Dialysis
B. Electrodialysis
C. Ultrafiltration
D. Decantation
E. Compensatory dialysis

133. Neuroleptanalgesia has been applied in the case of cardiac infarction. What neuroleptic is most often applied along with fentanyl?

A. Droperidol
B. Perphenazine (Aethaperazinum)
C. Levomepromazine
D. Clozapine
E. Sulpiride

134. A patient complaining of dry mouth, photophobia, and visual impairment has been delivered into an admission room. The skin is hyperemic and dry; pupils are dilated; tachycardia is observed. The patient was diagnosed with belladonna alkaloids intoxication. What drug would be advisable?

A. Proserin
B. Aceclidine
C. Pilocarpine
D. Armin
E. Dipiroxim

135. A patient with signs of cardiac glycosides intoxication was prescribed Unithiol. What is the mechanism of drug action in this case?

A. Reactivation of membrane $K^+$, $Na^+$-adenosine triphosphatase
B. Binding of ionized $Ca^{2+}$
C. Increased permeability of $K^+$ into myocardiocytes
D. Increased $Na^+$ content in myocardium
E. Induction of cardiac glycoside metabolism

136. A woman suffering from neurosis has disturbed sleep. What drug is optimal for insomnia treatment?

A. Nitrazepam
B. Phenobarbital
C. Aethaminalum-natrium (Pentobarbital)
D. Bromisoval
E. Valerian tincture

137. Select the halogenated antiseptic that would be preferable for a child to pack in the first aid kit, when going to a summer camp:

A. Iodine alcoholic solution
B. Brilliant green
C. Copper sulfate
D. Methylene blue
E. Formaldehyde solution

138. A patient consulted an ophthalmologist about deterioration of twilight vision and xerophthalmus. What drug should the doctor prescribe?
A. Retinol
B. Pyridoxine
C. Tocopherol
D. Ascorbic acid
E. Cocarboxylase

139. Due to prolonged taking of phenobarbital an epileptic patient has developed tolerance for this drug. What is this phenomenon based on?

A. Biotransformation acceleration
B. Absorption process weakening
C. Increase of receptor sensitivity
D. Biotransformation suppression
E. Substance accumulation in body

140. Decreased absorption of tetracyclines, if they are taken simultaneously with antacids, is an example of their:

A. Pharmacokinetic incompatibility
B. Pharmaceutical incompatibility
C. Pharmacodynamic incompatibility
D. Synergism
E. Functional antagonism

141. A patient with epilepsy is prescribed a diuretic. Name this drug:

A. Diacarb (Acetazolamide)
B. Verospiron
C. Furosemide
D. Hypothiazid (Hydrochlorothiazide)
E. Mannitol

142. What side effect is characteristic of captopril?

A. Dry cough
B. Increased blood pressure
C. Hyperglycemia
D. Cardiac rate disorder
E. Hypokalemia

143. Paracetamol belongs to the following pharmacological group:

A. Nonnarcotic analgetics
B. Soporifics
C. Diuretics
D. Hypotensive drugs
E. Antianginal drugs

144. An elderly patient suffers from constipation caused by large intestine hypotonia. What drug should be prescribed?

A. Bisacodyl
B. Sodium sulfate
C. Castor oil
D. Atropine sulphate
E. Procainamide

145. A 25-year-old woman with signs of acute morphine intoxication was administered naloxone, which rapidly improved her condition. What is the mechanism of action of this drug?

A. Opioid receptor blockade
B. GABA receptor blockade
C. Serotonin receptor blockade
D. Dopamine receptor blockade
E. Benzodiazepine receptor blockade

146. A patient with rheumatoid arthritis and concomitant duodenal ulcer has to be prescribed nonsteroid anti-inflammatory drug. Which one of the drugs listed below is a drug of choice in the given case?

A. Celecoxib
B. Acetylsalicylic acid
C. Paracetamol
D. Metamizole
E. Diclofenac sodium

147. A 48-year-old patient has been intravenously administered prednisolone solution to arrest severe attack of bronchial asthma. What group of hormonal agents does prednisolone belong to?

A. Glucocorticoids
B. Gestagenic drugs
C. Estrogenic drugs
D. Mineralocorticoid
E. Anabolic steroids

148. Choose the most efficient way of convallariae glycoside administration for acute heart failure treatment:
149. What drug should be administered in case of acute cardiac insufficiency?
A. Corglycon
B. Salbutamol
C. Pilocarpine hydrochloride
D. Naloxone
E. Heparin

150. A patient with signs of mercury poisoning has been delivered into an admission room. What antidote should be prescribed in this case?
A. Unithiol
B. Atropine sulfate
C. Proserin
D. Naloxone
E. Calcium chloride

151. Diuretic should be prescribed to treat cerebral edema. What drug is to be administered?
A. Furosemide
B. Hydrochlorothiazide
C. Caffeine and sodium benzoate
D. Diacarb (Acetazolamide)
E. Spironolactone

152. A doctor has prescribed a nonsteroidal anti-inflammatory drug to relieve inflammation and pain syndrome. Name this drug:
A. Diclofenac sodium
B. Glibenclamide
C. Loratadine
D. Prednisolone
E. Calcium chloride

153. The 55-year-old patient has been diagnosed with angina pectoris. Calcium channel-blocking agent was prescribed for treatment. Name this agent:
A. Amlodipine
B. Atenolol
C. Guanethidine
D. Reserpine
E. Labetalol

154. Colored or white component of double perianth, which consists of petals, is a:
A. Corolla
B. Flower cup
C. Androecium
D. Gynoecium
E. Perigonium

155. Weeds can be harmful for populace’s wellbeing. Particularly, allergic reactions are often caused by the following plant in its period of blossoming:
A. Ambrosia artemisiifolia
B. Equisetum arvense
C. Stellaria media
D. Erigeron canadensis
E. Taraxacum officinale

156. Aurococcus culture was obtained from the nasal cavity of a child suffering from chronic tonsillitis. Causative agent’s sensitivity towards a number of antibiotics was tested to choose the optimal drug. What drug WAS NOT included in antibiotic susceptibility testing?
A. Nystatin
B. Ampicillin
C. Tetracycline
D. Levomycetin (Chloramphenicol)
E. Erythromycin

157. Interferons are natural antiviral and antitumor agents. What is their mechanism of action?
A. Protein synthesis depression
B. Protein synthesis increase
C. Replication activation
D. Transcription activation
E. Repair activation

158. A patient of a neurology unit suffers from paralysis of all limbs. Name this condition:
A. Tetraplegia  
B. Paraplegia  
C. Hemiplegia  
D. Paresis  
E. Hypodynamia

159. The most severe and dangerous complication of diabetes mellitus is hypoglycemic coma that is characterized by loss of consciousness and is lethal, unless efficient emergency treatment is received by patient. What is the main pathogenetic component of hypoglycemic coma?

A. Carbohydrate deficiency and low energy of cerebral neurons  
B. Carbohydrate deficiency and low energy of myocardium cells  
C. Blood hyperosmia  
D. Noncompensated ketoacidosis  
E. Respiratory alkalosis

160. A patient demonstrates symmetrical dermatitis on the palms. A doctor made a diagnosis of pellagra. What vitamin deficiency can result in such symptoms?

A. Nicotinic acid  
B. Cobalamin  
C. Ascorbic acid  
D. Folic acid  
E. Cholecalciferol

161. A ready-made drug was inoculated on Sabouraud’s agar and incubated under 22°C for 5 days. This nutrient medium was used to determine the following:

A. Number of mold and yeast fungi  
B. Total number of bacteria  
C. Presence of *E. coli*  
D. Presence of *S. aureus*  
E. Presence of *Salmonella*

162. Specify the reagent allowing to determine barium cations in the presence of calcium and strontium cations:

A. Potassium dichromate  
B. Potassium chloride  
C. Potassium iodide  
D. Potassium nitrate  
E. Sodium hydroxide

163. The second stage of detoxification involves joining certain chemical compounds with functional groups of toxines. Select one such compound:

A. Glucuronic acid  
B. Higher fatty acids  
C. Cholesterol  
D. Glucose  
E. Pyruvate

164. What forms from an ovule after fertilization of flowering plants?

A. Seed  
B. Gametophyte  
C. Sporophyte  
D. Fruit  
E. Endosperm

165. A woman complains of nausea, vomiting, skin itch. She was diagnosed with mechanical jaundice. What is the possible cause of skin itch in such a condition?

A. Bile acids accumulating in the blood  
B. Increased blood content of indirect bilirubin  
C. Cholesterol accumulating in the blood  
D. Direct bilirubin appearing in the blood  
E. Erythrocyte disintegration products accumulating in the blood

166. When determining oxidizing agents by means of iodometry in the presence of starch the following phenomenon can be observed at the titration end point:

A. Blue coloring disappears  
B. Red coloring appears  
C. Green coloring of precipitate appears  
D. Green coloring of solution disappears  
E. White precipitate occurs
Corolla of a zygomorphic bi-sexual flower consists of 5 petals: the largest one is called banner, two lateral - wings, and two fused together - keel. This corolla is characteristic of Fabacea family and is called:

A. Papilionaceous  
B. Lingulate  
C. Rotate  
D. Funnelform  
E. Tubular

During ultrasound investigation a patient has been diagnosed with bilateral stenosis of renal artery with atherosclerotic genesis. Specify the bioactive substance that due to its excessive secretion is the key component of arterial hypertension pathogenesis in the given case:

A. Renin  
B. Cortisol  
C. Vasopressin  
D. Noradrenaline  
E. Thyroxin

A plant producing essential oil has square stem, bilabiate corolla, coenobium fruit. These features are characteristic of:

A. Lamiaceae  
B. Papaveraceae  
C. Polygonaceae  
D. Solanaceae  
E. Scrophulariaceae

Blood contains erythrocytes with sizes of $10^{-6}$ m degree as its constituent parts. What type of disperse system is blood?

A. Microheterogeneous  
B. Homogeneous  
C. Coarse dispersion  
D. Colloidal dispersion  
E. Heterogeneous

A patient with headache consulted a pharmacist. The patient was prescribed a cyclooxygenase inhibitor - an aminophenol derivative. What drug was prescribed?

A. Paracetamol  
B. Acetylsalicylic acid  
C. Diclofenac  
D. Ketorolac  
E. Ibuprofen

A woman is to be prescribed a narcotic analgesic for labor pain relief. What drug is indicated in this case?

A. Promedol (Trimeperidine)  
B. Morphine  
C. Papaveretum (Omnopon)  
D. Codeine  
E. Fentanyl

A woman, who during the 5th-10th weeks of her pregnancy had been taking sodium valproate for treatment of epilepsy, gave birth to a child with pathology of the vertebral column (split spine). What side effect of the drug caused such malformation?

A. Teratogenic  
B. Mutagenic  
C. Embryotoxic  
D. Fetotoxic  
E. Sensitizing

Analysis of a dry substance always begins with preliminary tests. Sample under investigation is green in color, which allows to conclude the presence of:

A. Chrome (III)  
B. Manganese (II)  
C. Cobalt (II)  
D. Iron (III)  
E. Barium (II)

Students should identify the following to determine the sex of a flower:

A. Stamens and pistils  
B. Flower cup and corolla  
C. Pedicle and receptacle  
D. Symmetry  
E. Color and type of indumentum

Disintegration of adenosine nucleotides results in release of ammonia. What enzyme plays the key
role in ammonia synthesis from these compounds?

A. Adenosine deaminase  
B. Alcohol dehydrogenase  
C. Lactate dehydrogenase  
D. Alanine transaminase  
E. Amylase

**177.** An analytical chemist performs qualitative analysis of cations of the II analytical group. The following solution is used to separate silver and mercury chlorides:

A. Ammonia  
B. Hydrochloric acid  
C. Sodium hydroxide  
D. Sodium nitrate  
E. Potassium chloride

**178.** Both scientific and folk medicine uses medicinal plant *Glycyrrhiza glabra L.* What part of the plant is harvested?

A. Roots and rhizomes  
B. Foliage  
C. Inflorescence  
D. Grass  
E. Seeds

**179.** A patient undergoes chemotherapy with 5-fluorouracil that is a competitive inhibitor of thymidilate synthase. What process is inhibited by this drug?

A. Thymidine monophosphate synthesis  
B. Purine nucleotides disintegration  
C. Adenosine triphosphate synthesis  
D. Purine nucleotides salvage  
E. Glucose synthesis

**180.** Upon increase of pressure the system’s chemical equilibrium will shift towards parent substances. Point out such a system:

A. \( N_2O_4(gas) \leftrightarrow 2NO_2(gas) \)  
B. \( C_{(solid)} + O_2(gas) \leftrightarrow CO_2(gas) \)  
C. \( 4HCl_{(gas)} + O_2 \leftrightarrow 2H_2O_{(gas)} + 2Cl_2(gas) \)  
D. \( N_2(gas) + 3H_2(gas) \leftrightarrow 2NH_3(gas) \)  
E. \( CO_2(gas) + H_2(gas) \leftrightarrow CO_{(gas)} + H_2O_{(gas)} \)

**181.** What substance is not a surfactant at the water solution-air interface?

A. Sodium chloride  
B. Acetic acid  
C. Ethyl alcohol  
D. Sodium stearate  
E. Glucose

**182.** A drop of oil-water emulsion had been applied to a plate covered with paraffin; no wetting was observed. Such phenomenon characterizes this emulsion as:

A. Direct  
B. Concentrated  
C. Diluted  
D. Stable  
E. Invert

**183.** A patient has been receiving Theophylline (inhibitor of cyclic adenosine monophosphate phosphodiesterase) for a week. What hormone can increase its action due to such treatment and cause hyperglycemia?

A. Glucagon  
B. Testosterone  
C. Aldosterone  
D. Insulin  
E. Estradiol

**184.** Burner’s flame colors carmine-red in the presence of salts of an unknown cation. Name this cation:

A. Strontium  
B. Ammonium  
C. Sodium  
D. Potassium  
E. Iron

**185.** Reaction of sodium ions with potassium hexahydroxoantimonate
(V) in neutral medium produces precipitate. Specify the color of this precipitate:

A. White  
B. Red  
C. Yellow  
D. Green  
E. Blue  

186. Rhizome of an *Asteraceae* family species is polyccephalous, succulent, has lysigenous cavities, accumulates inulin. Such underground organ is characteristic of:

A. *Inula helenium*  
B. *Hyoscyamus niger*  
C. *Digitalis grandiflora*  
D. *Sorbus aucuparia*  
E. *Helianthus annuus*  

187. Osmotic pressure is an important characteristic of biological fluids. Semipermeable membranes are necessary for penetration of solvent molecules. What substance *CANNOT* be used as a semipermeable membrane?

A. Glass  
B. Biological membrane  
C. Collodion film  
D. Parchment  
E. Gelatine  

188. Modelling of immobilization stress is performed on a test animal - guinea pig - that starved for a day. Dissection revealed hyperemic gastric mucosa with multiple erosions. What theory of ulcer formation is confirmed by this test?

A. Corticovisceral (stress)  
B. Vascular  
C. Inflammatory  
D. Mechanical  
E. Peptic  

189. What group of drugs is characterized by development of drug addiction as a side effect?

A. Psychosedatives  
B. Cholinergic antagonists  
C. Adrenergic drugs  
D. Diuretics  
E. Emetics  

190. What cation can be detected with Chugaiev’s agent (Dimethylglyoxime)?

A. $\text{Ni}^{2+}$  
B. $\text{Ca}^{2+}$  
C. $\text{K}^+$  
D. $\text{Mn}^{2+}$  
E. $\text{Co}^{2+}$  

191. A substance performs mechanical function; its cells are covered with uniformly thick lignified membranes. This substance is:

A. Sclerenchyma  
B. Collenchyma  
C. Periderm  
D. Cambium  
E. Sieve tubes  

192. A patient in the state of ketoacidotic coma presents with loud rapid respiration: labored expiration with tension of expiratory muscles occurs after deep inspiration. Name the type of pathologic respiration:

A. Kussmaul’s  
B. Cheyne-Stokes’  
C. Gasping  
D. Stenotic  
E. Biot’s  

193. Air contamination with pathological microorganisms can be anticipated by the presence of indicator bacteria. Specify the bacteria that indicate immediate epidemiologic danger:

A. Hemolytic streptococci  
B. Sarcinae  
C. Mold fungi  
D. Yeast fungi  
E. Micrococci  

194. Barium carbonate sol has been obtained from the reaction of excessive amount of barium chloride solution with ammonium carbonate solution-
on. The micelle aggregate of obtained sol consists of the following microcrystals:

A. \( BaCO_3 \)
B. \( NH_4Cl \)
C. \((NH_4)_2CO_3\)
D. \( BaCl_2 \)
E. Mixture of \( BaCO_3 \) with \( NH_4Cl \)

195. A 40-year-old man presents with rapid weight gain after he had suffered a severe craniocerebral trauma. On examination the patient’s weight was 125 kg, with his height being 175 cm. What mechanism of obesity development is the most likely in this case?

A. Hypothalamic
B. Alimentary
C. Hormonal
D. Hereditary
E. -

196. Different structures of a bacterial cell perform different functions. What dispensable component of a cell ensures its survival within hostile environment?

A. Spores
B. Flagella
C. Capsule
D. Cilia
E. Inclusions

197. Isoniazid

\[
\begin{array}{c}
\text{O} \\
\text{N} \\
\text{H} \\
\text{N-H}_2 \\
\end{array}
\]

is an antituberculous drug derivative from:

A. Isonicotinic acid
B. Nicotinic acid
C. Picolinic acid
D. Pyrrole-2-carboxylic acid
E. N-aminobenzoic acid

198. What substance will react with propane under the given conditions?

A. \( Br_2, h\nu, 20^oC \)
B. \( H_2SO_4 \) concentrated
C. \( HNO_3 \) concentrated
D. \( Cl_2, FeCl_3 \)
E. \( CH_3COONO_2 \)

199. Specify the name of the carbohydrate given in the figure

\[
\begin{array}{c}
\text{CH}_2=\text{CH}-\text{CH}_2-\text{CH}_2-\text{CH}_2-\text{C}-\text{CH}_3 \\
\text{CH}_3 \\
\end{array}
\]

given according to the systematic nomenclature:

A. 5,6,6-trimethyl-1-hepten
B. 2,2,3-trimethyl-6-hepten
C. 5-tert-Butyl-1-hexene
D. 2-tert-Butyl-5-hexene
E. 2,2,3-trimethyl-6-hexene

200. What reaction proves that phenol has acidic properties?

A. 

\[
\begin{array}{c}
\text{O} \\
\text{H} \\
\end{array} + NaOH \rightarrow \begin{array}{c}
\text{O} \\
\text{Na}^+ \\
\end{array} + H_2O
\]

B. 

\[
\begin{array}{c}
\text{O} \\
\text{H} \\
\end{array} + 3\text{Br}_2 \rightarrow \begin{array}{c}
\text{Br} \\
\text{Br} \\
\text{Br} \\
\end{array} + 3\text{HBr}
\]

C. 

\[
\begin{array}{c}
\text{O} \\
\text{C} \\
\text{C} \\
\text{CH}_3 \\
\end{array} + (CH_2CO)_2 \rightarrow \begin{array}{c}
\text{O} \\
\text{C} \\
\text{C} \\
\text{CH}_3 \\
\end{array} + CH_3COOH
\]

D. 

\[
\begin{array}{c}
\text{O} \\
\text{H} \\
\end{array} + 2\text{Br}_2 \rightarrow \begin{array}{c}
\text{Br} \\
\text{Br} \\
\text{Br} \\
\text{Br} \\
\end{array} + \text{HBr} + 2\text{HBr}
\]

E. 

\[
\begin{array}{c}
\text{O} \\
\text{H} \\
\end{array} + \text{H}_2 \rightarrow \begin{array}{c}
\text{H} \\
\text{H} \\
\end{array}
\]


INSTRUCTIONAL BOOK

Testing Board

TEST ITEMS FOR LICENSING EXAMINATION: KROK 1. PHARMACY.

Kyiv. Testing Board.
(English language).

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<table>
<thead>
<tr>
<th>Abbreviation</th>
<th>Description</th>
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<tbody>
<tr>
<td>A/G</td>
<td>Albumin/globulin ratio</td>
</tr>
<tr>
<td>A-ANON</td>
<td>Alcoholics anonymous</td>
</tr>
<tr>
<td>ACT</td>
<td>Abdominal computed tomography</td>
</tr>
<tr>
<td>ALT</td>
<td>Alanin aminotranspherase</td>
</tr>
<tr>
<td>AP</td>
<td>Arterial (blood) pressure</td>
</tr>
<tr>
<td>AST</td>
<td>Aspartat aminotranspherase</td>
</tr>
<tr>
<td>BP</td>
<td>Blood (arterial) pressure</td>
</tr>
<tr>
<td>BR</td>
<td>Breathing rate</td>
</tr>
<tr>
<td>bpm</td>
<td>Beats per minute</td>
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<tr>
<td>C.I.</td>
<td>Color Index</td>
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<tr>
<td>CBC</td>
<td>Complete blood count</td>
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<tr>
<td>CHF</td>
<td>Chronic heart failure</td>
</tr>
<tr>
<td>CT</td>
<td>Computer tomography</td>
</tr>
<tr>
<td>DIC</td>
<td>Disseminated intravascular coagulation</td>
</tr>
<tr>
<td>DCC</td>
<td>Doctoral controlling committee</td>
</tr>
<tr>
<td>DM-2</td>
<td>Non-Insulin dependent diabetes mellitus</td>
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<tr>
<td>DTP</td>
<td>Anti diphtheria-tetanus vaccine</td>
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<tr>
<td>ECG</td>
<td>Electrocardiogram</td>
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<td>ESR</td>
<td>Erythrocyte sedimentation rate</td>
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<td>FC</td>
<td>Function class</td>
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<td>FEGDS</td>
<td>Fibro-esphago-gastro-duodenoscopy</td>
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<tr>
<td>Gy</td>
<td>Gray</td>
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<tr>
<td>GIT</td>
<td>Gastrointestinal tract</td>
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<tr>
<td>Hb</td>
<td>Hemoglobin</td>
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<tr>
<td>HbA1c</td>
<td>Glycosylated hemoglobin</td>
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<tr>
<td>Hct, Ht</td>
<td>Hematocrit</td>
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<tr>
<td>HDL</td>
<td>High-density lipoproteins</td>
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<tr>
<td>IDDM</td>
<td>Insulin dependent diabetes mellitus</td>
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<td>IHD</td>
<td>Ischemic heart disease</td>
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<tr>
<td>IU</td>
<td>International unit</td>
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<td>MSEC</td>
<td>Medical and sanitary expert committee</td>
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<td>NIDDM</td>
<td>Non-Insulin dependent diabetes mellitus</td>
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<tr>
<td>pCO₂</td>
<td>CO₂ partial pressure</td>
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<tr>
<td>pO₂</td>
<td>O₂ partial pressure</td>
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<tr>
<td>Pm</td>
<td>Per minute</td>
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<td>Ps</td>
<td>Pulse rate</td>
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<td>R</td>
<td>Roentgen</td>
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<td>RBC</td>
<td>Red blood count</td>
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<td>Rh</td>
<td>Rhesus</td>
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<tr>
<td>RR</td>
<td>Respiratory rate</td>
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<td>S&lt;sub&gt;1&lt;/sub&gt; (S&lt;sub&gt;1&lt;/sub&gt;)</td>
<td>Heart sound 1</td>
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<tr>
<td>S&lt;sub&gt;2&lt;/sub&gt; (S&lt;sub&gt;2&lt;/sub&gt;)</td>
<td>Heart sound 2</td>
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<td>TU</td>
<td>Tuberculin unit</td>
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<td>U</td>
<td>Unit</td>
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<td>USI</td>
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<td>V/f</td>
<td>Vision field</td>
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<tr>
<td>WBC</td>
<td>White blood count</td>
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<tr>
<td>X-ray</td>
<td>Roentgenogram</td>
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